

MASS BALANCE ANALYSIS OF REACTIONS AND REACTORS

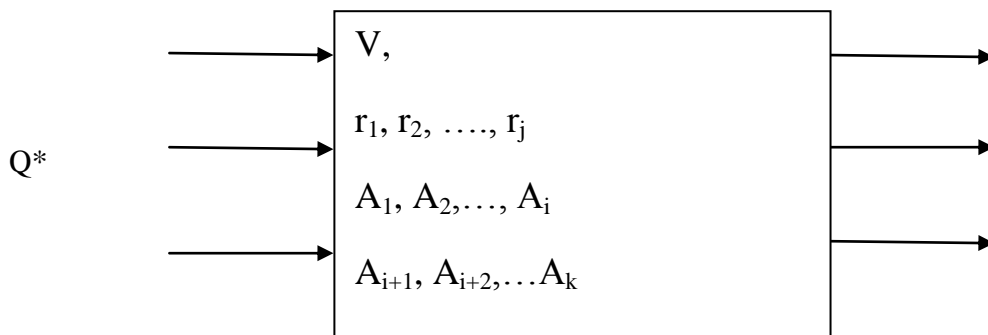
1. General mass balance concept

$$\text{Mass}_i \text{ inflow} - \text{Mass}_i \text{ outflow} + \text{Mass}_i \text{ generation} = \text{Mass}_i \text{ accumulation}$$

2. Multiple mass balances required for process that have coupled reactions, with overlapping reactants and products

- Oxidation of organic matter and oxygen uptake
- Heterotrophic cell synthesis uptake of organic matter
- Cell synthesis uptake of nitrogen
- Oxidation of ammonia and autotrophic cell synthesis
- Reduction of nitrate and simultaneous consumption of organic matter
- Lysis of cells and generation of degradable organic matter
- And others...

Mass balance schematic for reactor volume, V , with multiple reactions r_j , with inflow/outflow rate, Q , multiple reactants A_i , and multiple products A_k is appropriate for biological wastewater treatment reactor, where we will keep track of COD, N, O_2 , cells, etc.



3. Two elements of mass balances:

- Components (products and reactants)
- Rates (kinetics)

For component A_1 , mass balance on mixed volume with a single reaction rate, r_1 :

$$Q^*A_{1,0} - Q^*A_{1,e} + V^*r_1 = V \quad \text{---}$$

Where V = reactor volume (m^3), Q = volumetric flow rate (m^3/d), $A_{1,0}$ = influent concentration of A_1 , $A_{1,e}$ = effluent concentration of A_1 , and A_1 = reactor concentration (g/m^3).

NOTE: one mass balance per component. For the system above, could write k mass balances each with $1 \rightarrow j$ reaction terms.

4. **Stoichiometric analysis and coefficients** enables us to express consumption and production of a reactant or product in terms of other components in the reactor system which is necessary to evaluate multiple reactions
 - a. For example, calculating the oxygen requirement for consumption of inflows of COD and ammonia nitrogen require the stoichiometric coefficients for oxygen in both reactions
5. **Kinetic (rate) expressions** are second element of mass balances required to predict effluent and reactor component concentrations.