

Anaerobic biological treatment for removal of inorganic contaminants from drinking water

Workshop on Biological Drinking Water Treatment
IWA Leading Edge Technology Conference
June 1, 2010

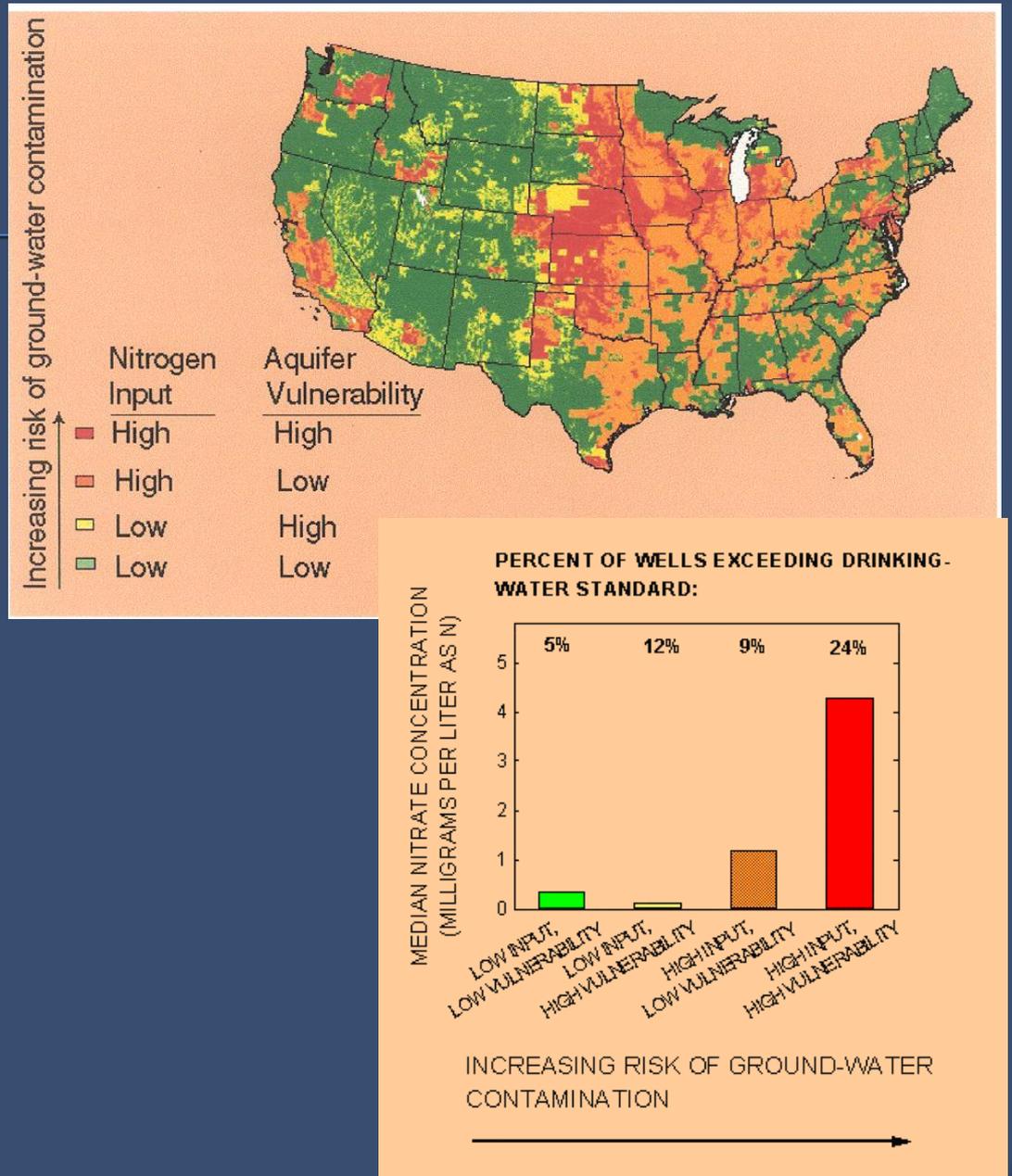
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○ Nitrate risk factors

- Nitrate loading from agriculture, livestock operations
- Aquifer vulnerability: sandy formation, shallow wells

○ Affected communities

- Small town or rural
- Inland

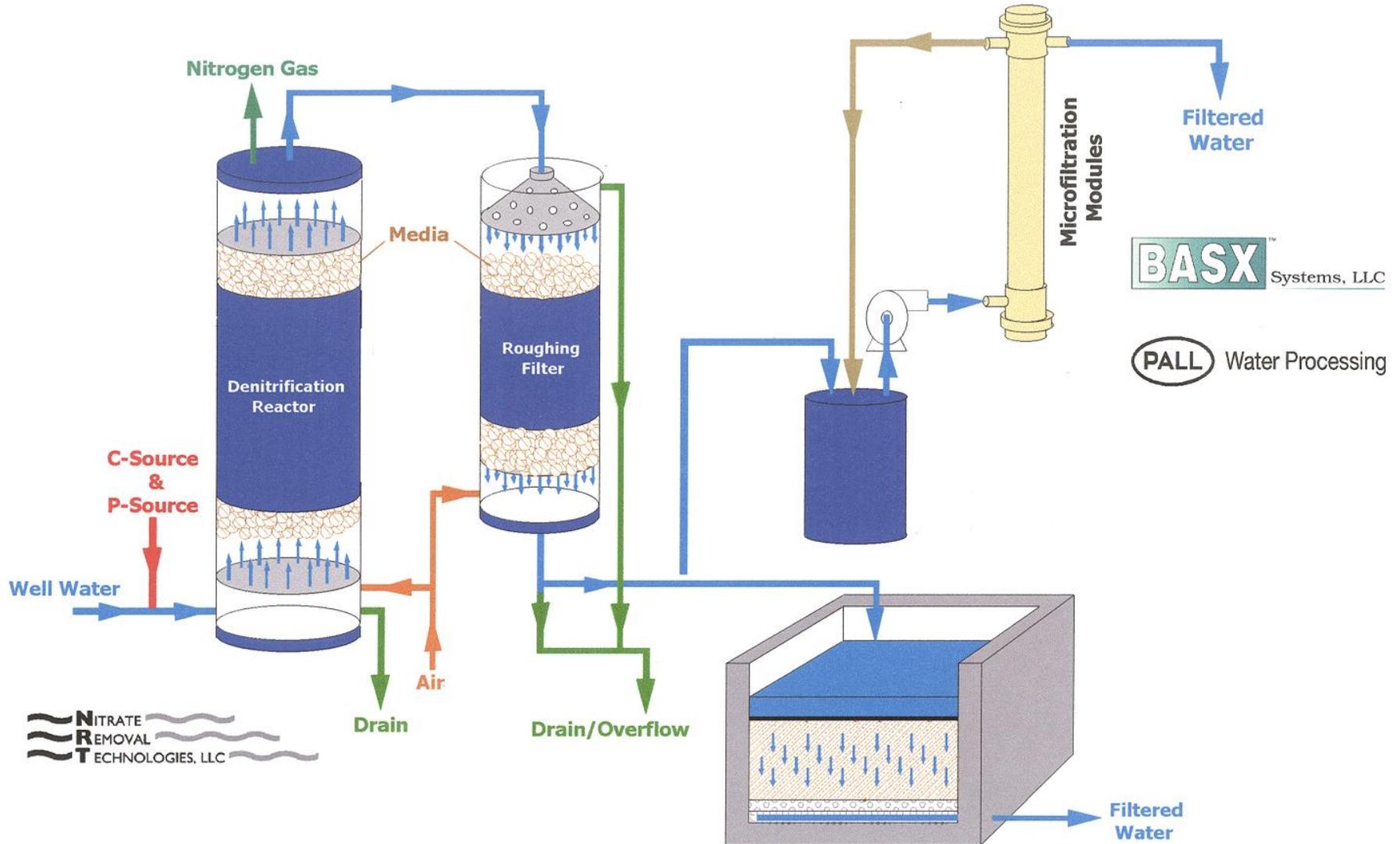


Nolan et al., USGS, 1997 survey of 1400 wells

Objectives of drinking water denitrification studies

- Production of finished drinking water in field conditions and scale to meet regulatory and utility interests
- Demonstrate operability in small-utility settings
- Provide unit cost estimates of biological denitrification process
- Study nitrate-perchlorate competition for substrate

Denitrification-Post-Treatment Process Flow Diagram



Plant Characteristics

Location	Purpose	Configuration	Flow (lpm)	Influent NO ₃ (mg/L-N)	Duration (days)
Wiggins	Demo.	Denitrify - SSF	38	15-25	365 (180 data)
Suffolk Mt. Sinai	Demo.	Denitrify - SSF, MF	30 (DN) 8 (MF, SSF)	9	243 (151 data)
Coyle	Drinking Water ¹	Denitrify -SSF	106 (cap), 53 (DN)	10	758 (24 data)

¹ 290 residents, 400 schoolchildren

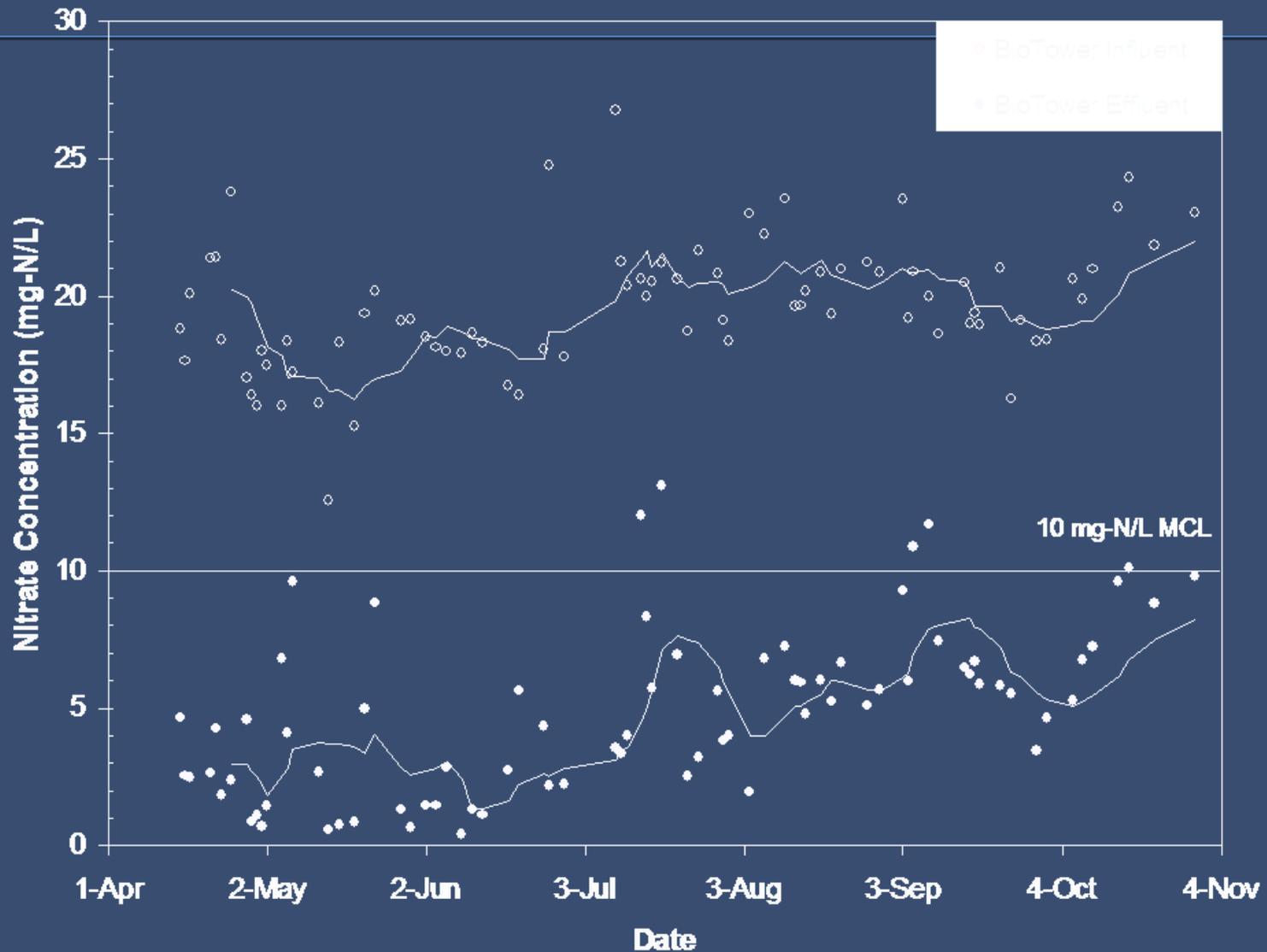
Wiggins, Colorado Well Pump House (left) and
Demonstration Plant Building (right)



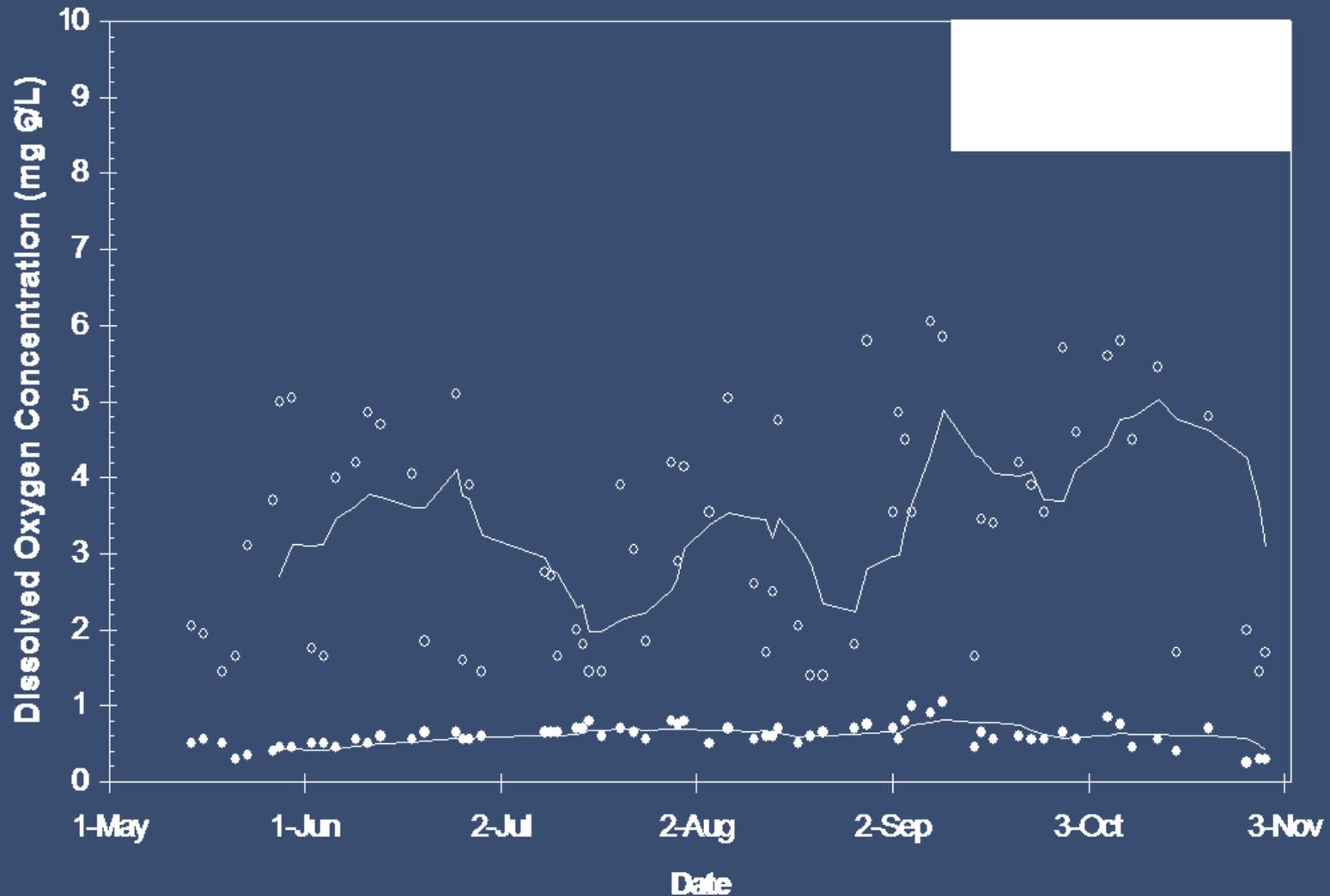
Wiggins demonstration plant: (right) anoxic denitrifying biotowers and aerobic prefilter; (left) slow sand filter.



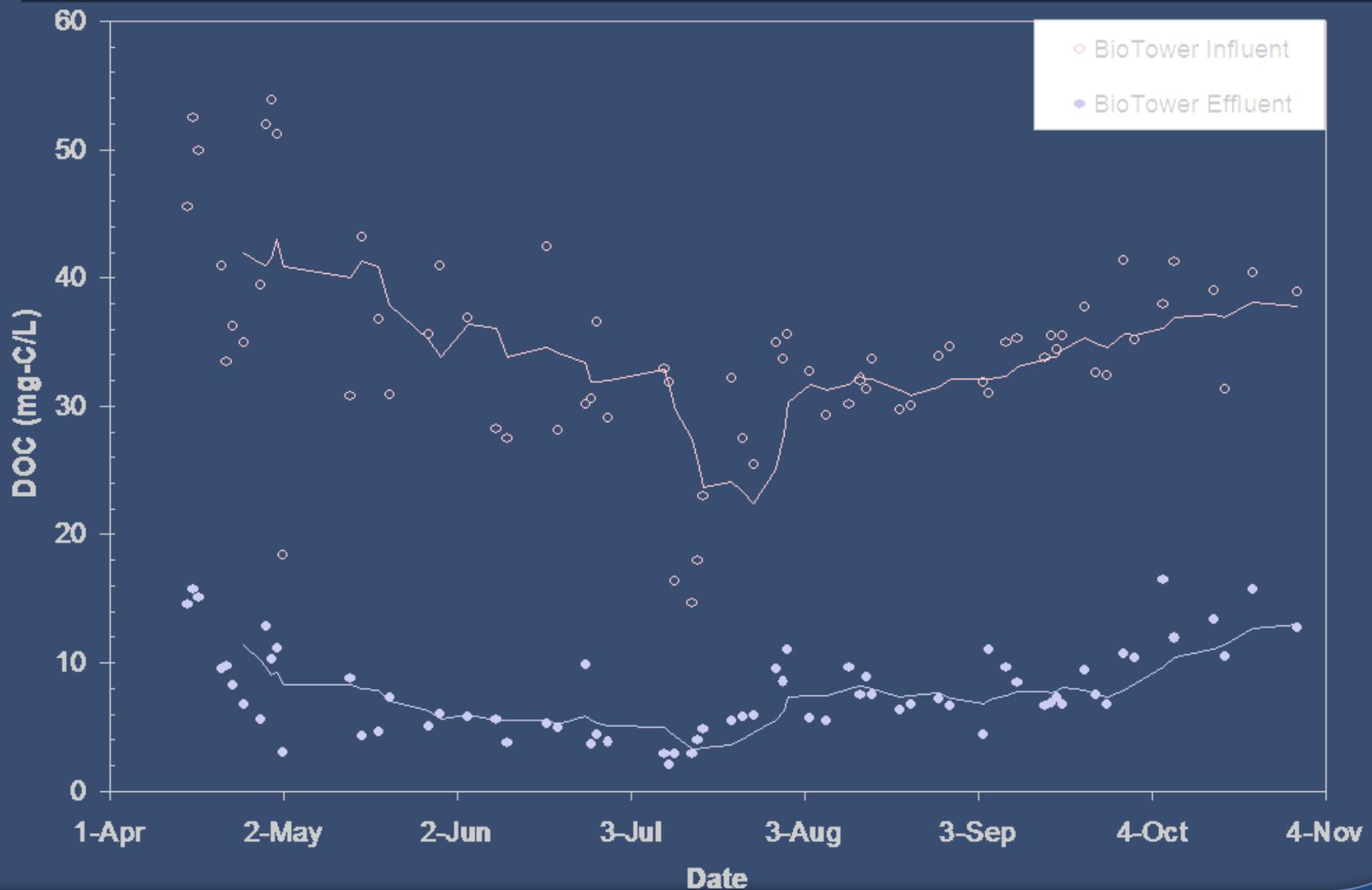
Wiggins biotower influent and effluent nitrate (lines are 14-day moving averages)



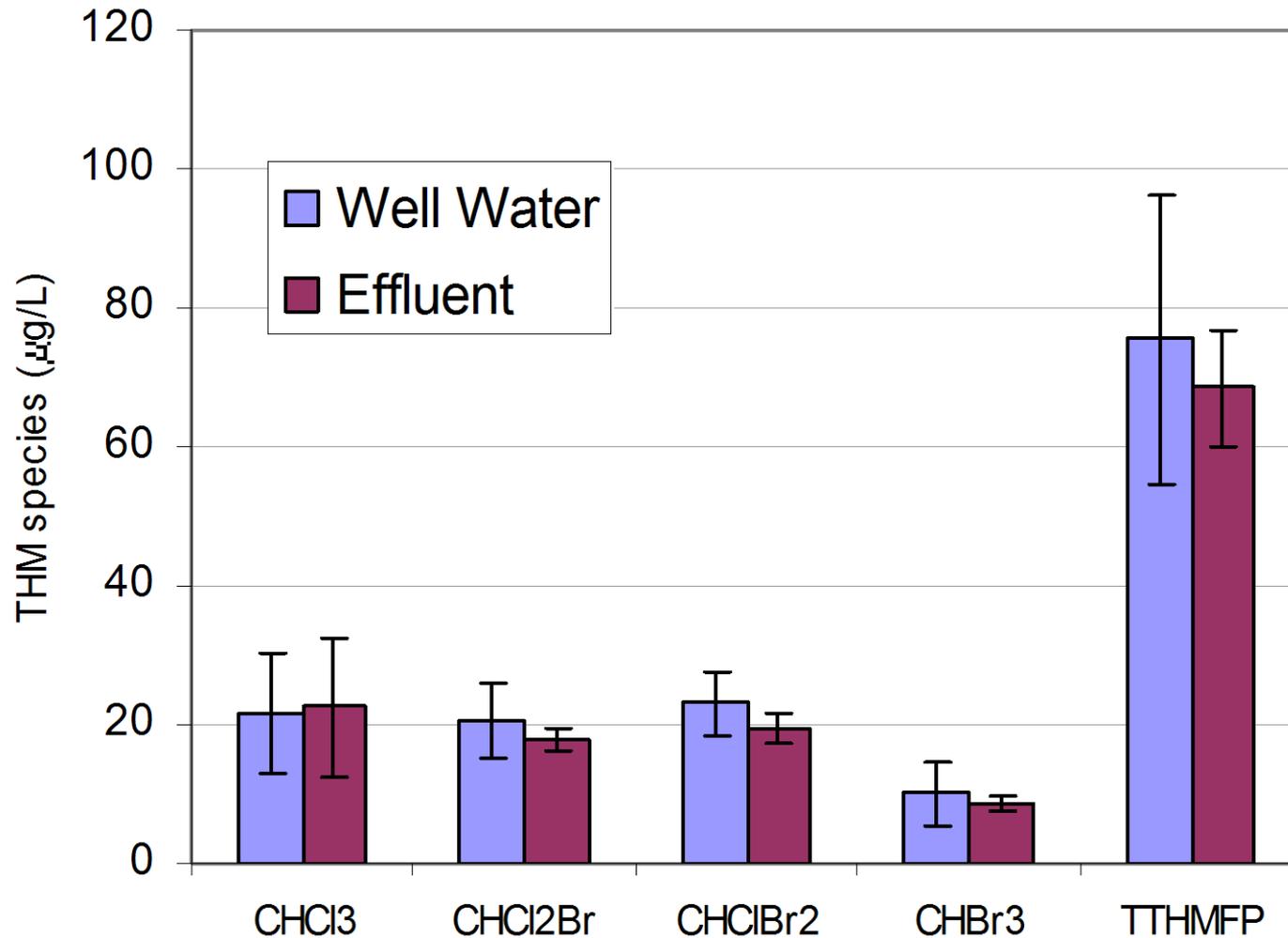
Biotower Influent and Effluent Dissolved Oxygen at Wiggins Plant (lines are 14-day moving averages)



Influent (after corn syrup addition) and effluent TOC at Wiggins plant (lines are 14-day moving averages)



Individual Species and Total Trihalomethane Formation Potential in Wiggins Plant Influent and Effluent



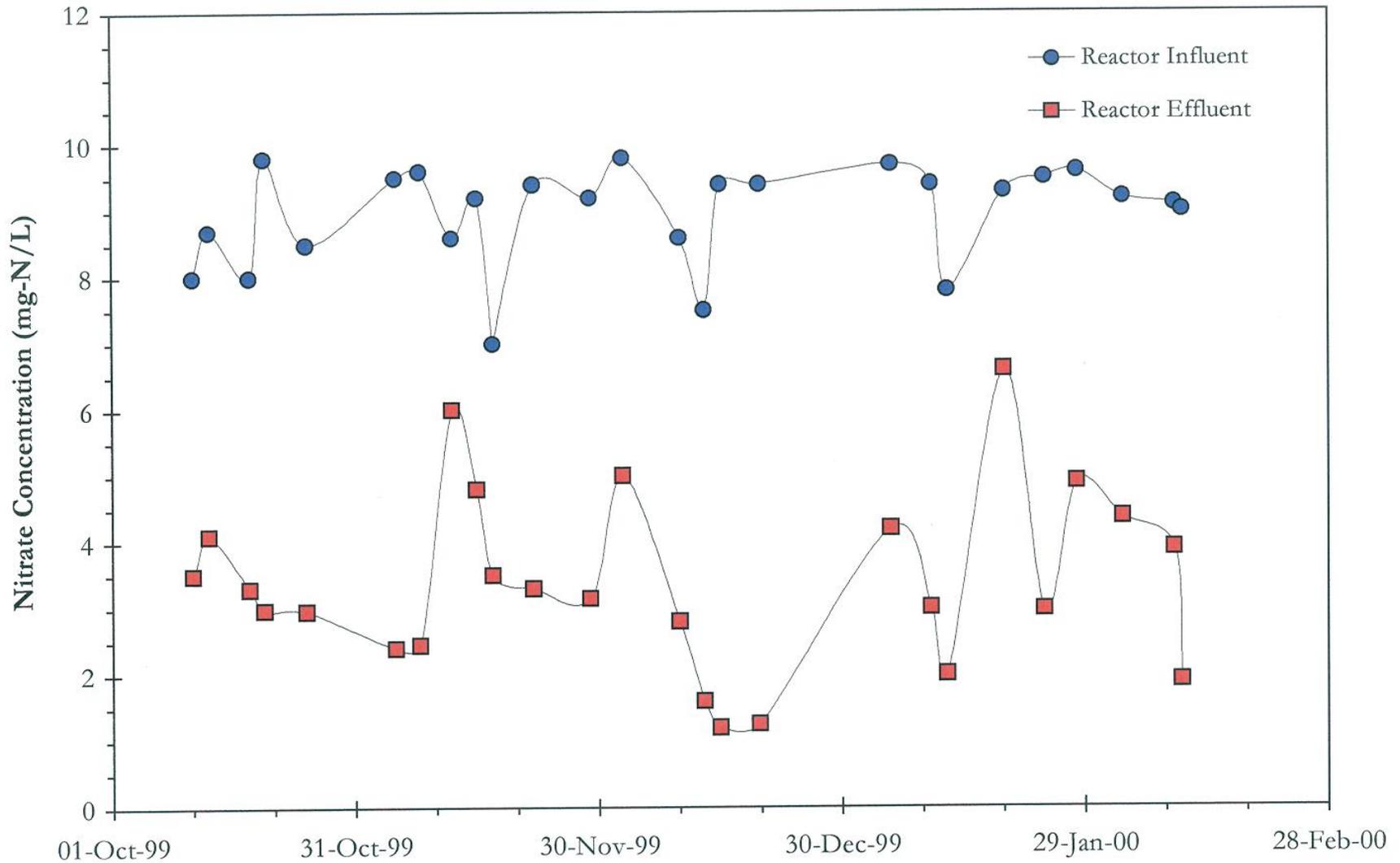
Summary of Wiggins Demonstration

- Consistent denitrification achieved from average 19.6 to 4.3 mg/L N @ C:N = 1.3:1
- Influent dissolved oxygen 2 – 6 mg/L did not affect performance
- No significant change in DBP precursors
- Significant increase in Cl-demand from 0.6 to 4.5 ppm
- Corn syrup associated with high biomass growth, TOC, clogging, high effluent coliform and HPC bacteria, and NH₄ in effluent

Left: ceramic microfiltration system (BASX Systems). Right: hollow fiber membrane microfiltration system (Pall Corp.) used at Suffolk Demonstration Plant



Biotower Influent and Effluent Nitrate at Suffolk Plant (acetate carbon addition to achieve 3 mg/L-N in effluent)



Biotower Comparison

Parameter	Wiggins	Mt. Sinai
Carbon source	Corn Syrup	Acetate
Nitrate Removal Rate (kg-N/m ³ -media/d)	0.34	0.40
Effluent Nitrite (mg/L-N)	1.4	0.04
Effluent TOC (mg/L)	7.7	6
Effluent Turbidity (NTU)	2.52	0.9
Effluent Total Coliform (cfu/100 mL)	5.1×10^6	65
HPC (cfu/100mL)	6.6×10^7	2.4×10^5

Suffolk Filter Comparison

Parameter	Slow Sand Filter	Ceramic Microfilter	Hollow Fiber Memb. MF
Effluent TOC (mg/L)	0.4	1.5	0.4
Effluent Turbidity (NTU)	0.3	0.27	0.03
Effluent Total Coliform (cfu/100 mL)	1	ND	ND
Effluent HPC (cfu/100 mL)	2,900	170	1
Chlorine Demand (mg/L)	0.55	0	0

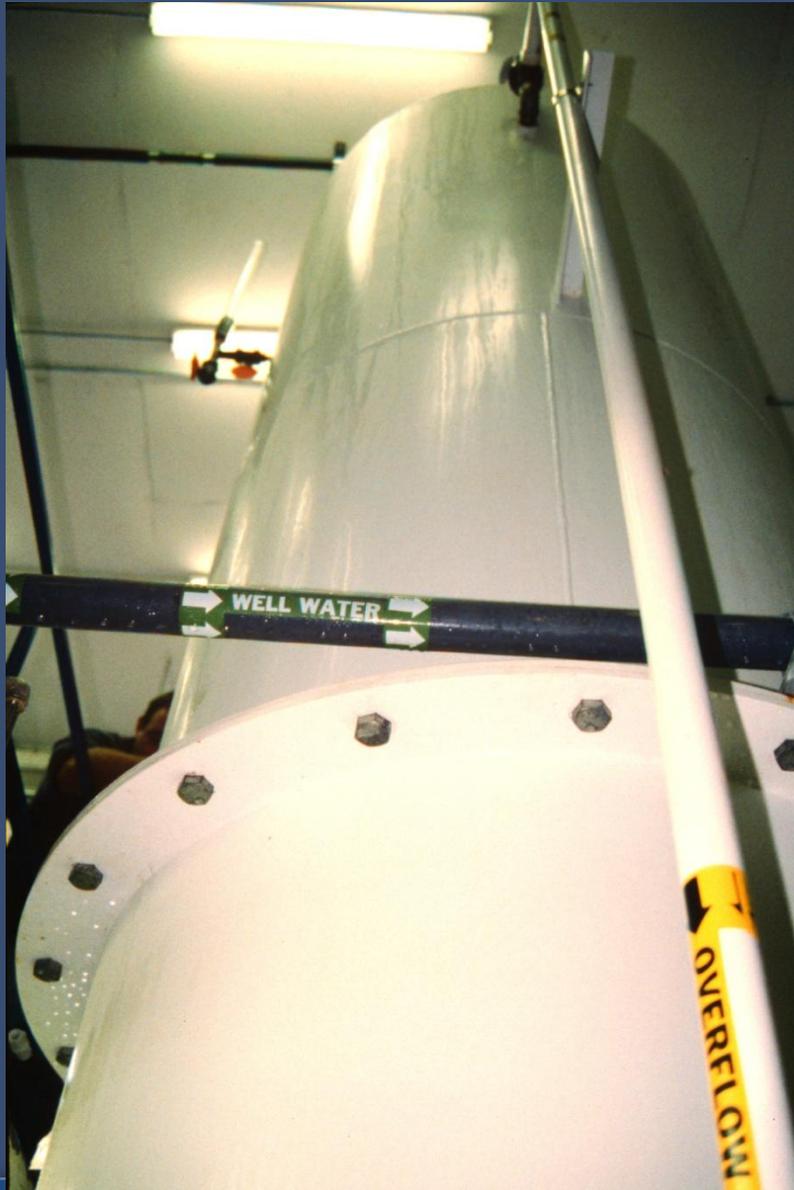
Suffolk Study Summary

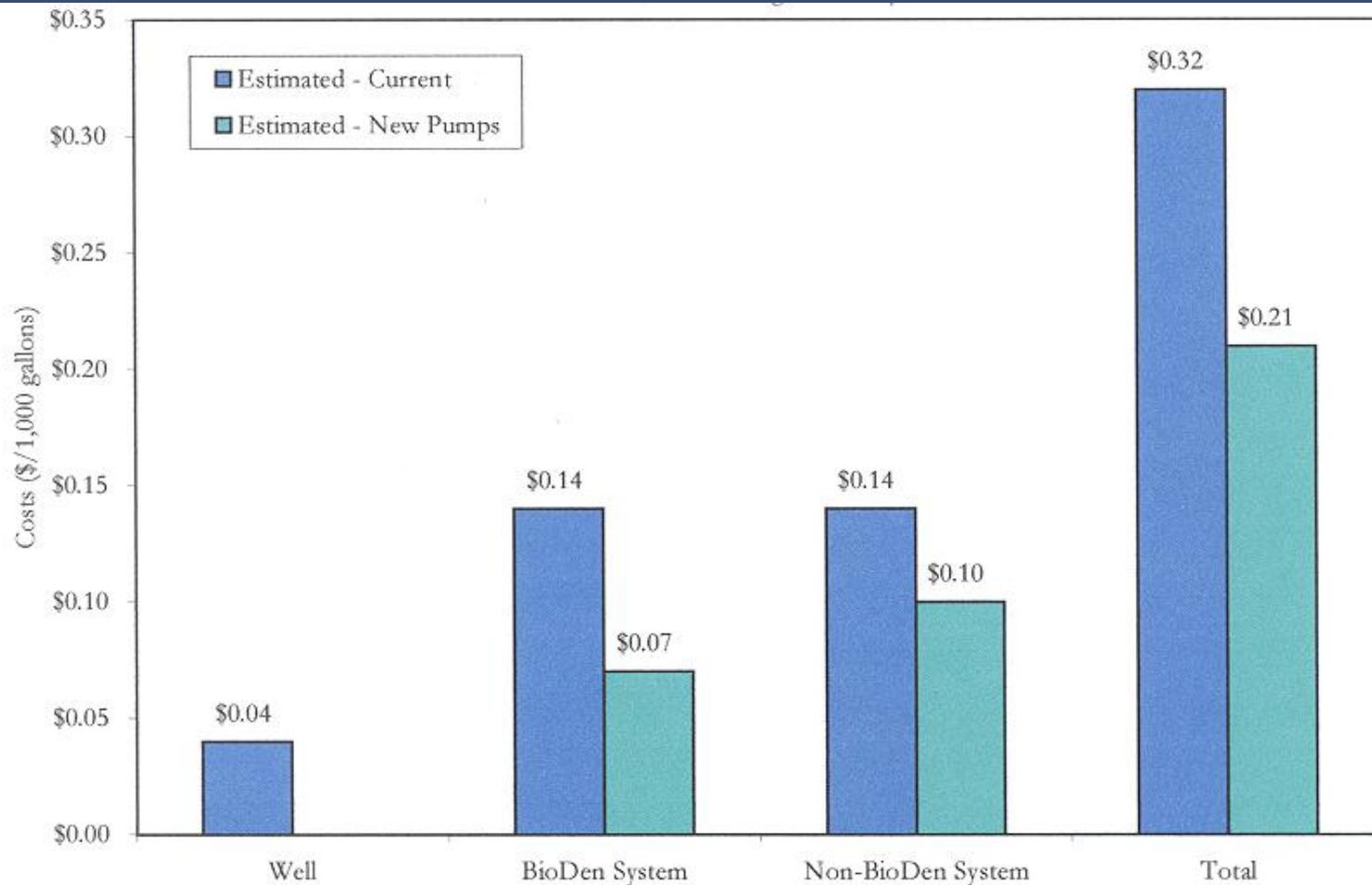
- Use of acetate as carbon source reduces biotower effluent bacteria, turbidity and nitrite.
- Microfiltration produces bacteria-free low turbidity product water
- 90% of DOC was removed in SSF and hollow fiber MF
- Residuals from MF = 3% of product water
- Chlorine injection to prevent fouling may account for rise in THMFP after MF treatment

Coyle, Oklahoma Drinking Water Denitrification Plant



Town of Coyle, Oklahoma Drinking Water Treatment Plant



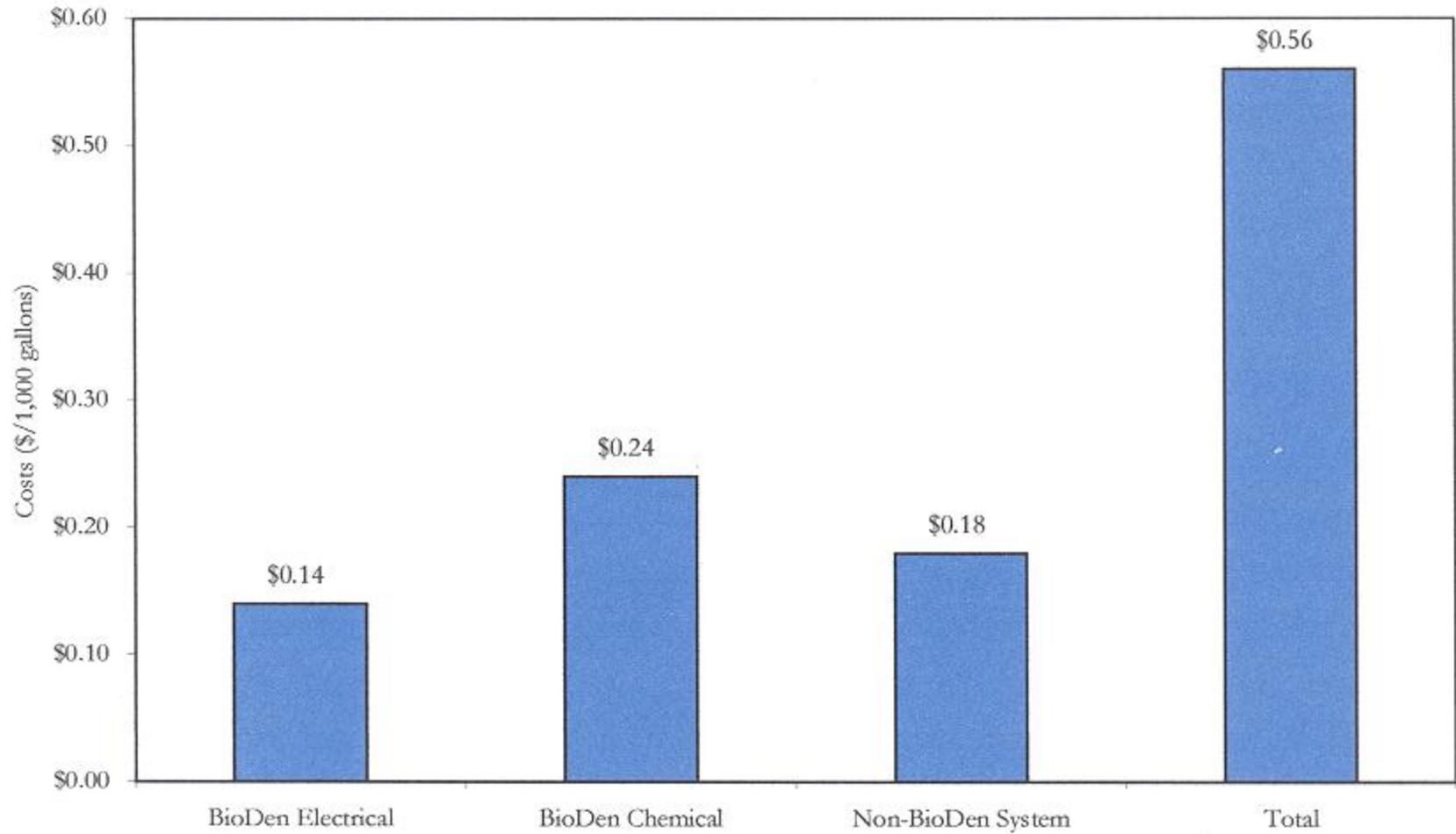


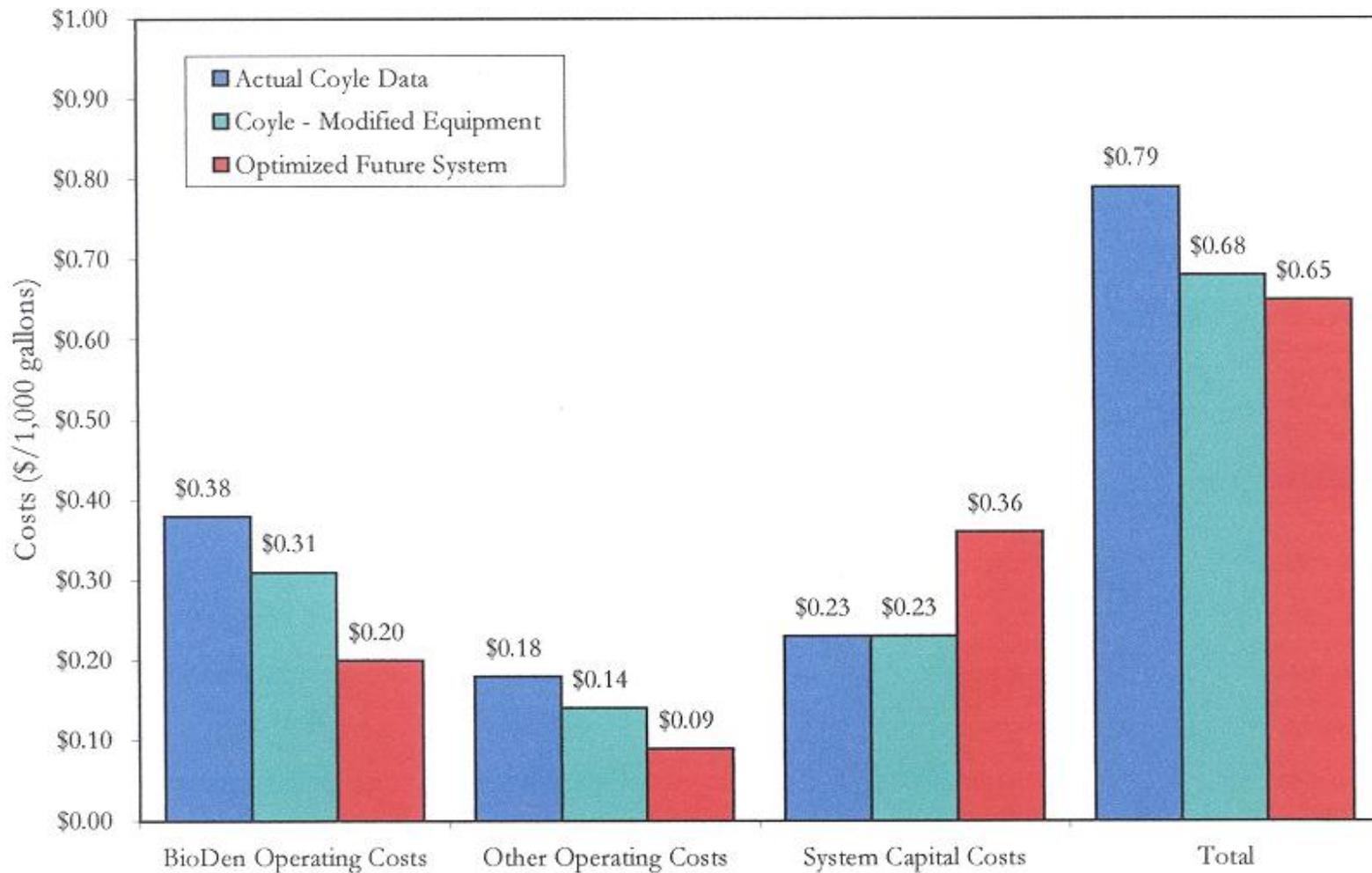
Power Cost at Coyle Denitrification Plant
 Unit electric power cost: \$0.0532/kwh

Chemical Costs

Chemical	Before Blending (/1,000 gallons)	After Blending (/1,000 gallons)
Vinegar (acetate)	\$0.36	\$0.18
Phosphate	\$0.10	\$0.05
Other (analytical)	\$0.02	\$0.01
Total	\$0.48	\$0.24

Total Operating Costs at Coyle Plant





TOTAL COSTS

Usage: 15 million gallons per year (28 gpm)

Coyle plant: \$130,000 capital (67% grant), 20 years, 5%

Future system: \$130,000 capital, no grant, 20 years, 5%

Cost Study Summary

- Total unit cost (capital, O & M) for drinking water treatment (including well pumping) was \$0.79/1,000 gallons (with 67% capital financing grant from State of Oklahoma)
- Chemical costs are approximately 30% of total. Carbon (vinegar) is 75% of chemical cost.
- Power costs are approximately 40% of treatment total
- Estimated unit cost with future system of equal capacity assuming pump replacement and plant configuration changes, and no grants is \$0.65/1,000 gallons

Acknowledgements

Nitrate Removal Technologies (NRT)

National Water Research Institute

EPRI/NRECA

Colorado Dept. Local Affairs

Town of Brighton, CO, Town of Wiggins, CO

Suffolk County Water Authority (SCWA),

Town of Mt. Sinai, New York , Town of Coyle,
OK

BASX

Pall Corp.

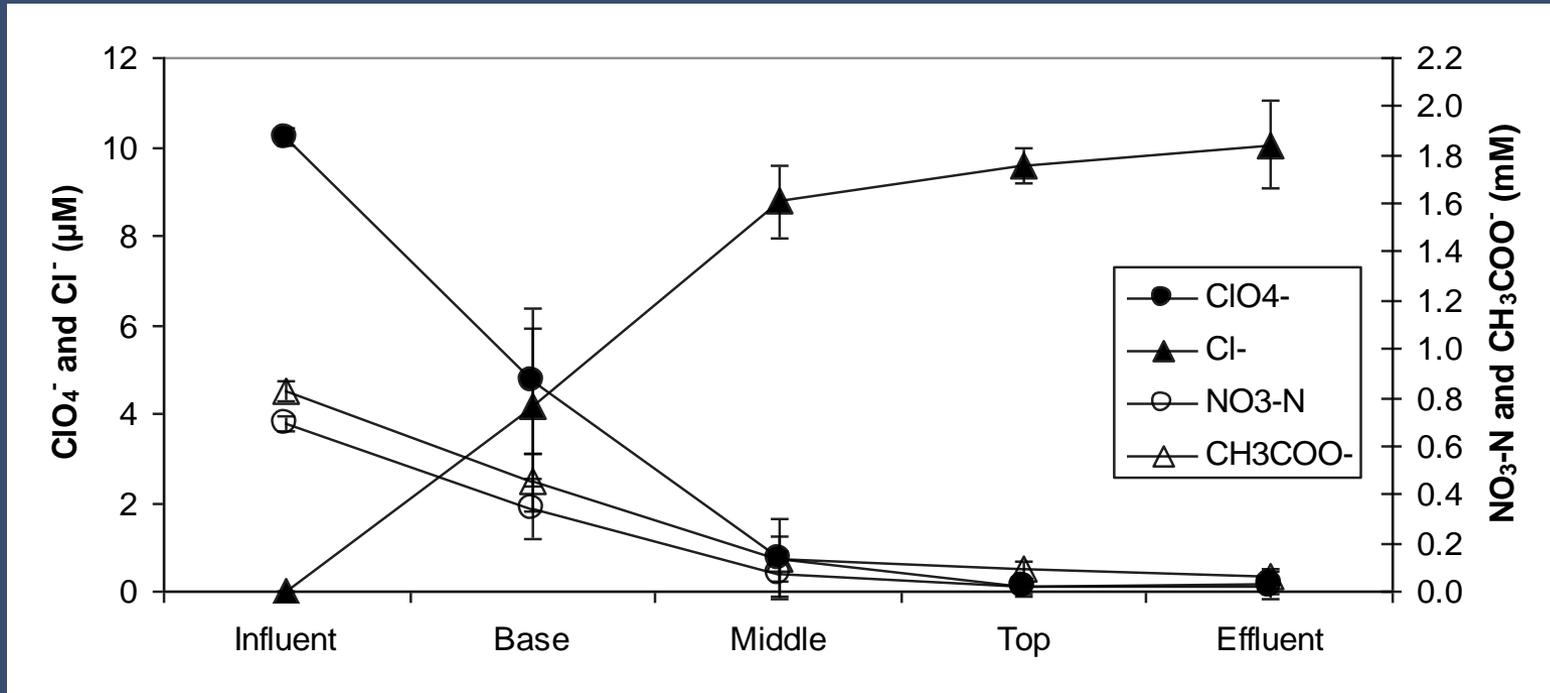
Colorado Dept. Public Health & Environment

Oklahoma Dept. Environmental Quality

Biological reduction of mixed nitrate and perchlorate influent in biofilm reactor

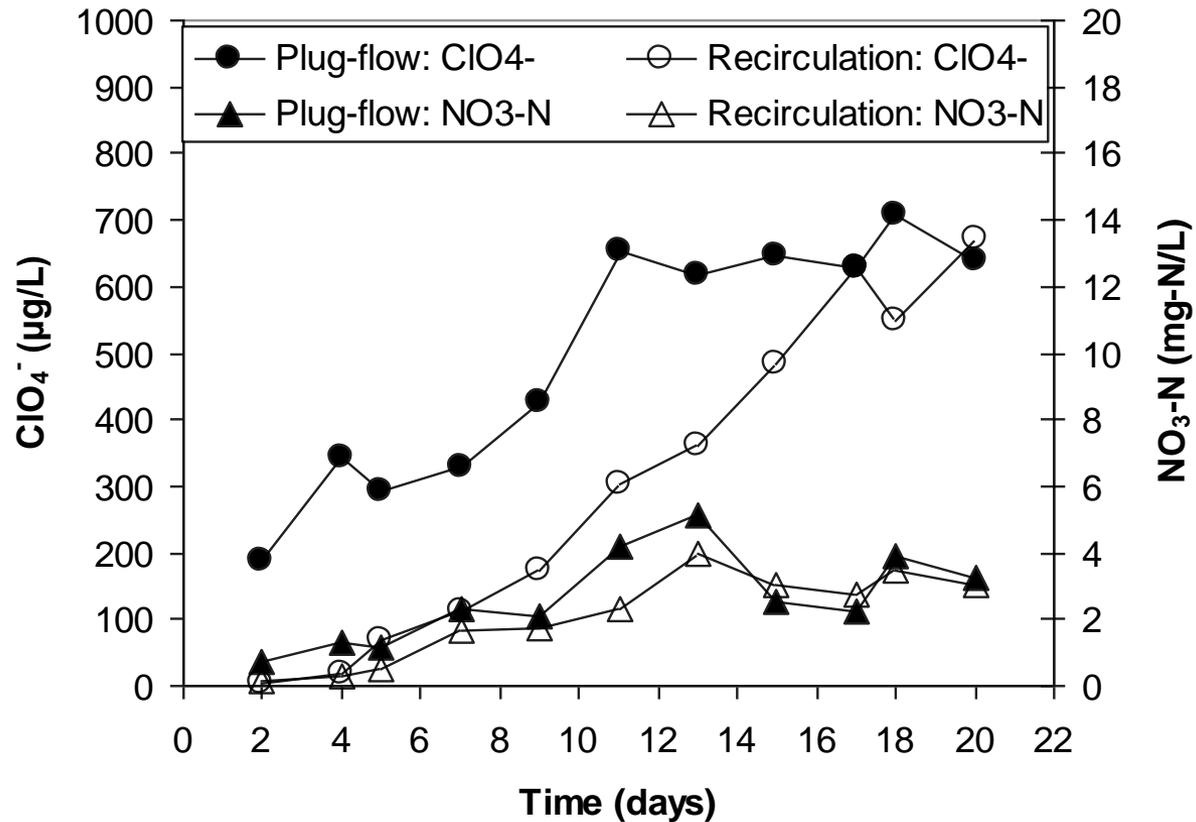


Biofilm reactor profiles. C:(N+ClO₄) = 1.8:1



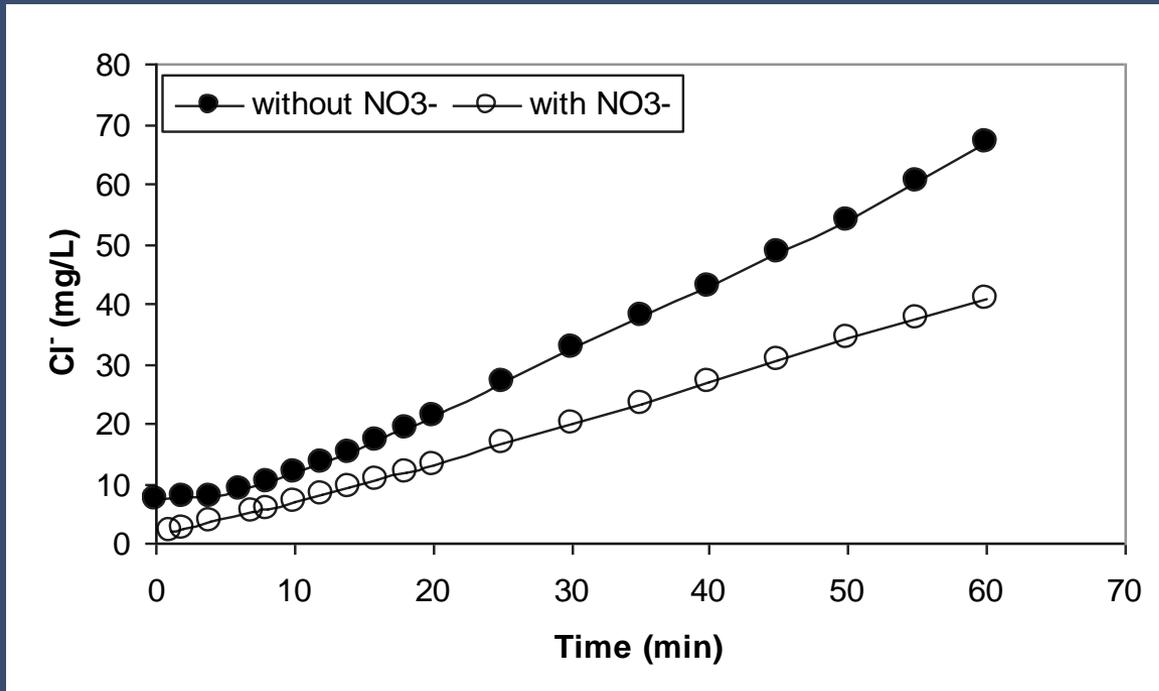
ClO₄⁻, NO₃⁻, CH₃COO⁻, and Cl⁻ concentration profiles in the biofilm reactor receiving 10 mg.L⁻¹ NO₃-N and 1,000 µg.L⁻¹ ClO₄⁻, and 42 mg.L⁻¹ CH₃COO⁻.

Biofilm reactor effluent. C:(N+ClO₄) = 1.2:1



Influent contained 1,000 µg.L⁻¹ ClO₄⁻, 16 mg.L⁻¹ NO₃-N and 52 mg.L⁻¹ acetate.

Effect of nitrate on chloride evolution in suspended cultures with excess acetate



Effect of 2 mM (28 mg/L) nitrate-N on reduction of 2 mM (200 mg/L) perchlorate in flasks inoculated with perchlorate-reducing culture, average MLSS in flasks was 5 g.L⁻¹. 16.7 mM acetate was added to both flasks. C:(ClO₄+N)>1.8

Biofilm reduction of mixed perchlorate-nitrate influent

- Perchlorate and nitrate are reduced simultaneously along bioreactor flow path when sufficient acetate added $C:(ClO_4+N) > 1.8$.
- Low substrate conditions $C:(ClO_4+N) < 1.2$ decreases perchlorate reduction more than denitrification
- Perchlorate-reducing cultures reduced nitrate without acclimation
- Nitrate competed with perchlorate for substrate electrons even when mass ratio of $ClO_4:N = 7$ g/g